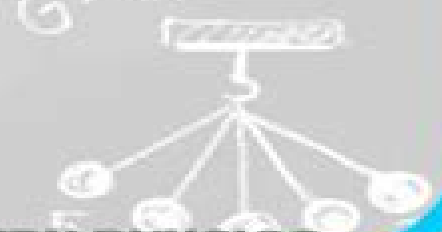
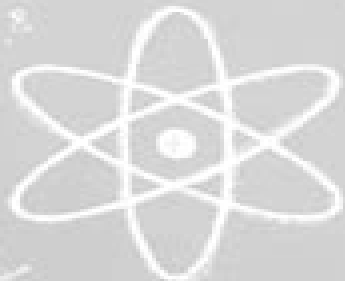


Belousov zhabotinsky reaction

- The Belousov-Zhabotinsky reaction is a classic example of non-equilibrium thermodynamics, showing how chemical oscillations can occur in a mixture
- This reaction demonstrates that a system can oscillate between different states instead of reaching a stable equilibrium
- It's commonly used to study chaos theory and chemical kinetics
- The reaction is visually striking, with colors that change periodically, making it a fascinating subject in both educational and research contexts

Practical example: Other areas

- The Belousov-Zhabotinsky reaction is a chemical oscillator, displaying colorful, rhythmic patterns in a liquid medium
- It's a captivating manifestation of non-equilibrium thermodynamics
- Although not encountered in everyday life directly, it helps us understand natural processes like heartbeats or circadian rhythms
- This reaction showcases how complex behaviors can emerge from simple chemical systems



The Belousov Zhabotinskii Reaction

Arturo Cuomo



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