Calculate the wavelength of the following radio waves? Circle the one that is of higher energy.

an FM radio wave at 106.7 MHz and an AM radio wave at 570 kHz.

2. Calculate the frequencies of the following electromagnetic radiation. Circle the one that is of higher energy. higher energy.

An IR-ray with $\lambda = 3.44 \times 10^{-4}$ m and visible light with $\lambda = 5.11 \times 10^{-7}$ m. 3.41 ×10-4 = 2.22 ×10, 200 = 2.22 ×10 = 2.22

Consider two beams of yellow light (which are in the visible region of the electromagnetic spectrum). One beam has its wavelength doubled, the other has its frequency doubled. Which beam of light is now in the UV region? increive E

4. A laser used in LASIK laser eye surgery has a wavelength of 193 nm. What is the frequency in hertz and the energy in joules of one photon of radiation of this laser? (E=hv), h = Planck's constant, 6.63 x 10⁻³⁴ J sec

35 = 2.958 × 108 De (1.55 × 1015 Az) = (663×10-34 × 10-18)

(b) A CO₂ laser used in dermatology for skin resurfacing operates at a frequency of v = 2,828 x 10⁷ s⁻¹. What is the wavelength of this electromagnetic radiation? What is the energy in joules of

Draw the Bohr atom model of the following:

Mihai V. Putz

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