

Standard Grade Chemistry  
Credit Level  
**Topic 3**  
Homework 2

1. (2011C)

A student made the following statements about the particles found in an atom.

A	Relative mass = 1
B	Charge = zero
C	Found outside the nucleus
D	Charge = 1+
E	Charge = 1-

Identify the **two** statements which apply to an electron. (1KU)

2. (2010C)

The grid shows information about some particles.

A $^{23}_{11}\text{Na}$	B $^{18}_8\text{O}$	C $^{40}_{19}\text{K}^+$
D $^{24}_{12}\text{Mg}^{2+}$	E $^{35}_{17}\text{Cl}^-$	F $^{16}_8\text{O}$

(a) Identify the **two** particles with the same number of neutrons. (1KU)

(b) Identify the particle which has the same electron arrangement as neon. (1PS)

3. (2009C)

The grid contains information about the particles found in atoms.

A relative mass = 1	B charge = zero	C relative mass almost zero
D charge = 1-	E found outside the nucleus	F charge = 1+

Identify the **two** terms which can be applied to protons. (2KU)

# Standard Grade Chemistry With Answers

**David Hawley, Sandy McLeman**



## **Standard Grade Chemistry With Answers:**

*Standard Grade Chemistry* Norman Conquest, Roddy Renfrew, 2002 The second edition of this title for standard grade chemistry matches the requirements of the standard grade and provides coverage of general and credit level material The authors have revised each chapter taking into account of new developments such as in the field of fossil fuels Other chapters have been updated where relevant for instance to reflect changes to the Relative Atomic Mass data The chemical dictionary chapter has been enlarged to include more key words New exam questions have been included together with homework and whole course questions at the end of the book

**Standard Grade Chemistry with Answers** Roddy Renfrew, Norman Conquest, 1995 This book has been specifically developed for the new Standard Grade Chemistry syllabus Building on the authors teaching experience the book addresses the real needs of teachers and students and provides a complete coherent course for all students tackling this course and examination

**Answers to Revision Questions for Standard Grade Chemistry** John M. Briggs, 2007

**Answers to Revision Questions for Standard Grade Chemistry** John M. Briggs, D. A. Buchanan, J. R. Melrose, 1998-03

**Standard Grade Chemistry** David Hawley, Sandy McLeman, 2012-07-25

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About the author s David is an experienced PT and Chemistry author He is also the SQA s Principal Examiner at Higher

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analytical process tools and computational methods and resources and to illustrate with problems that bring realism to the practice and importance of analytical chemistry It is designed for undergraduate college students majoring in chemistry and in fields related to chemistry New York Army National Guard, Education Services Handbook New York (State). Army National Guard,1986

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**Pamphlet - Dept. of the Army** United States. Department of the Army,1951

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samarium based reductants in organic and inorganic reactions mainly on those containing iodide and pentamethylcyclopentadienyl ligands Authoritative Comprehensive Up to date Critical Reliable Analytical Chemistry Refresher Manual John Kenkel, 2020-08-26 Analytical Chemistry Refresher Manual provides a comprehensive refresher in techniques and methodology of modern analytical chemistry Topics include sampling and sample preparation solution preparation and discussions of wet and instrumental methods of analysis spectrometric techniques of UV vis and IR spectroscopy NMR mass spectrometry and atomic spectrometry techniques analytical separations including liquid liquid extraction liquid solid extraction instrumental and non instrumental chromatography and electrophoresis and basic theory and instrument design concepts of gas chromatography and high performance liquid chromatography The manual also covers automation potentiometric and voltammetric techniques and the detection and accounting of laboratory errors Analytical Chemistry Refresher Manual will benefit all laboratory workers water and wastewater professionals and academic researchers who are looking for a readable reference covering the fundamentals of modern analytical chemistry

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