

Date Performed _____

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/20

Activity No.2 SPECIFIC HEAT OF SOLIDS

INTRODUCTION

Calorimetry means “measuring heat.” When heat flow occurs between two bodies that are isolated from their surroundings, the amount of heat lost by one body must be equal to the amount of heat gained by the other. Heat is a form of energy in transit, so this principle is just a conservation of energy. We take each quantity added to a body as positive and each quantity leaving a body as negative. When several bodies interact, the algebraic sum of the quantities of heat transferred to all the bodies must be zero.

The mathematical representation of the amount of heat Q needed to achieve a temperature change by a material with mass m and specific heat capacity c is

where T = final temperature, and initial temperature.

In an isolated system, there is no transfer of either mass or energy to its environment. For such case the total heat in the system is conserved, that is,

Specific Heat Capacity for Some Materials/Substance

Material/Substance	Specific Heat c	
	(cal/g $^{\circ}$ C)	(J/Kg $^{\circ}$ C)
Water	1.000	4190
Copper	0.093	390
Lead	0.031	130
Steel	0.118	470
Aluminum	0.217	910

OBJECTIVE

To measure the specific heat capacity of two kinds of solid specimens by calorimetry.

APPARATUS

Steam generator with cup, calorimeter, 2 thermometers, 2 kinds of metal shots, platform balance, slotted masses, 600 ml. beaker, electric stove, extension cord, rug

PROCEDURE

<https://www.youtube.com/watch?v=d4iCJakz2i8>

1. Fill the steam generator 1/2 full of water and start heating it over the stove. Set the stove range dial to **medium**.
2. Record the two kinds of metal shots in Table 2.1.
3. Weigh about 100 grams of the first metal shots. Pour these in the steam can (inner cup of the steam generator) and place the can in the steam generator. Cover the steam can with a cork with two holes, one for the thermometer and the other for the stirrer. Heat the shots over the steam generator and stir occasionally. **Do not use the thermometer to stir the shots** but keep the thermometer embedded in the shots. Continue heating and observe the thermometer until all the

Specific Heat Of Solids

Y. S. Touloukian



Specific Heat Of Solids:

Unveiling the Magic of Words: A Overview of "**Specific Heat Of Solids**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Specific Heat Of Solids**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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