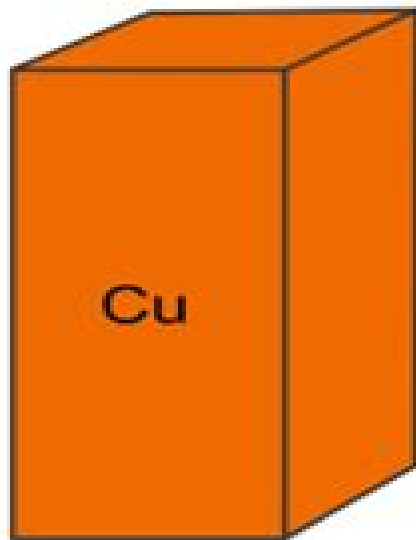
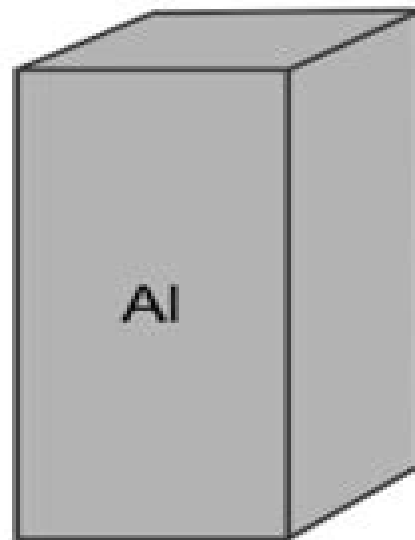




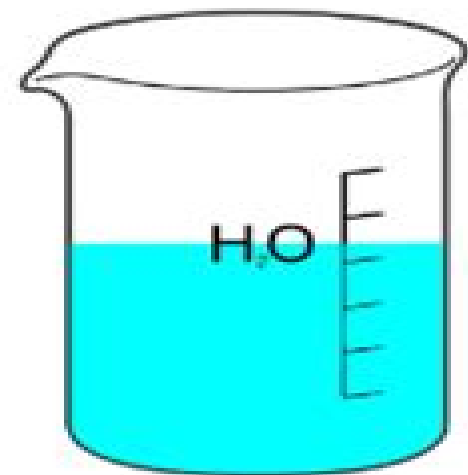
copper block

specific heat
of copper
 $= 390 \text{ J/kg}^\circ\text{C}$



aluminum block

specific heat
of aluminum
 $= 910 \text{ J/kg}^\circ\text{C}$



water

specific heat
of water
 $= 4200 \text{ J/kg}^\circ\text{C}$



Lower Specific Heat Capacity -

warms up and cools down quickly
as it takes much less energy to
change its temperature

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warms up and cools down slowly
as it takes up much more energy
to change its temperature

Specific Heat Of Solids

Edwin Irle Griggs



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