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# Sir Humphrey Gilbert's Voyage to Newfoundland

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examples are given in table 2 6 names of some ionic compounds table 2 6 compounds containing

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bismuth and the mass of tellurium per mole to find the total molar mass divide the mass of bismuth

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bonding is an important and extensive concept in chemistry and it will be treated in considerable

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it is ionic write the symbols for the ions involved a  $\text{NF}_3$  b  $\text{BaO}$  c  $\text{NH}_4\text{Cl}$  d  $\text{SrH}_2$  e  $\text{PO}_4^{3-}$  f  $\text{Na}_2\text{O}$  answer a answer b  
answer c answer d answer e answer f problem 4 3 1 4 4 3 1 4

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and nothing else so it is pure when we mix two different pure

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name a molecular compound to name a compound its the prefix name of first element prefix and name of the second element  
with the suffix ide the prefixes are as follows mono 1 di 2 tri 3 tetra 4

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substances as water  $\text{H}_2\text{O}$   $\text{H}_2\text{O}$  and carbon dioxide  $\text{CO}_2$   $\text{CO}_2$  these compounds are very different from ionic compounds like  
sodium chloride  $\text{NaCl}$   $\text{NaCl}$

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ch 2 ch 3 answer a answer b answer c answer d problem 4 4 1 3 calculate the molecular or formula mass of each of the following a p 4 b h 2 o c ca no 3 2 d ch 3 co 2 h acetic acid

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