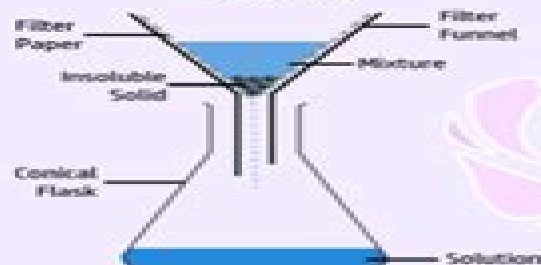


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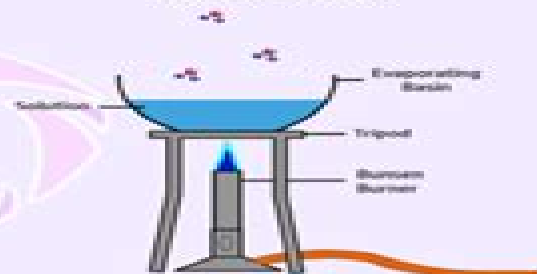
A. Filtration and crystallisation:

Filtration:



- ✧ Filtration is a technique used to separate a liquid from an **insoluble solid** – a solid that does not dissolve.
- ✧ The filter paper has tiny holes in it that allow water and dissolved solids to pass through.
- ✧ The holes are too small, however, for large particles to pass through – which means solids stay in the filter paper as **residue**.

Crystallisation:



- ✧ Crystallisation is a technique used to separate a liquid from a **soluble solid** – a solid that does dissolve.
- ✧ The Bunsen Burner is used to **evaporate** the water, leaving behind the **crystals** of your salt in the evaporating basin.
- ✧ After half of the water is evaporated, **leave it to cool**. The crystals can then be dried between filter paper.

B. Simple Distillation and Fractional Distillation:

Simple Distillation:



- ✧ Simple distillation is a technique used to separate two different **liquids** based on their **boiling points**.
- ✧ Example: Separating water (boils at 100°C) and ethanol (boils at 78°C).
 - ⚠ The solution is put into a round bottom flask and heated.
 - ⚠ Once it reaches 78°C, the ethanol will evaporate.
 - ⚠ The gas will rise and move into the **condenser**.
 - ⚠ The condenser is surrounded by cold water, which cools the gas down – turning it back into a **liquid**.
 - ⚠ If you stay **below** 100°C, the water will not evaporate.

Fractional Distillation:



- ✧ Fractional distillation is very similar to simple distillation, but is used to separate a mixture that contains lots of different liquids.
- ✧ All that is added is a **fractionating column**.
- ✧ As the mixture is heated, a **temperature gradient** will exist – where it is hotter at the bottom and cooler at the top.
- ✧ The vapour will rise up the column until it falls below its boiling point and will condense back down.
- ✧ Gases with a **lower boiling point** will be able to go **higher** up the column – and will be collected first and will go into the condenser.
- ✧ As the fractionating column continues to warm up, more and more of the fractions will move into the condenser, allowing them all to be collected separately.

Techniques Of Chemistry Vol 12 Separation And Purification

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Techniques Of Chemistry Vol 12 Separation And Purification:

Techniques of Chemistry, 1971 *Principles of Chemical Separations with Environmental Applications* Richard D. Noble, Patricia A. Terry, 2004-03-25 Chemical separations are of central importance in many areas of environmental science whether it is the clean up of polluted water or soil the treatment of discharge streams from chemical processes or modification of a specific process to decrease its environmental impact This book is an introduction to chemical separations focusing on their use in environmental applications The authors first discuss the general aspects of separation technology as a unit operation They also describe how property differences are used to generate separations the use of separating agents and the selection criteria for particular separation techniques The general approach for each technology is to present the chemical and or physical basis for the process and explain how to evaluate it for design and analysis The book contains many worked examples and homework problems It is an ideal textbook for undergraduate and graduate students taking courses on environmental separations or environmental engineering **Ion Exchange and Solvent Extraction** Jacob A. Marinsky, Yitzhak Marcus, 1987-10-29 This volume will capture transformational changes in both the chemistry and engineering side of solvent extraction creating new directions and deepening our understanding of the structure and dynamics of liquid liquid systems from the molecular to nano to meso to bulk scale Reviews will cover advances in microfluidics new tools for understanding the structure and dynamics of the liquid liquid interface ionic liquids in liquid liquid extraction molecular dynamics to visualize interactions in the solvent phase liquid liquid electrochemistry to interrogate the energetics of interfacial transport and complexation design of new extractants and the streamlining of process applications **Impediments to Analysis** J. E. Foster, 1980 **Techniques of Melt Crystallization** Gilbert J. Sloan, Andrew R. McGhie, 1988-01-18 Organic Solvents Physical Properties and Methods of Purification Fourth Edition Volume II in the *Techniques of Chemistry* series Edited by Arnold Weissberger Edited by John A Riddick William B Bunger and Theodore K Sakano This fourth edition updates and expands the material of the 15 year old third edition Besides revising the physical properties of preparation techniques for previously noted solvents over 150 new solvents have been added many selected to complete groups of isomers and expand homologous series listed in the previous edition Several isomeric and homologous series are initiated here including the xylenols chlorinated toluenes xylenes picolines lutidines and silanes 1986 0 471 08467 0 1 325 pp *Physical Methods of Chemistry Volume One Components of Scientific Instruments and Applications of Computers to Chemical Research* Second Edition Edited by Bryant W Rossiter and John F Hamilton Now in a revised Second Edition the classic *Physical Methods of Chemistry* series is an independent 8 volume set which surveys the most prevalent methods of determining a wide variety of physical properties of matter Volume One presents general laboratory techniques common to many of the specific physical methods that are detailed in subsequent volumes An excellent new work that also serves as a resource of information on a spectrum of components tools and techniques that are often used

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