

4. A rigid tank whose volume is unknown is divided into two parts by a partition, as shown in Fig. Q4. One side of the tank contains an ideal gas at 927°C . The other side is evacuated and its volume is twice that to the gas containing part. The partition is now removed and the gas quickly expands to fill the entire tank. Heat is now applied to the gas until the pressure equals the initial pressure.

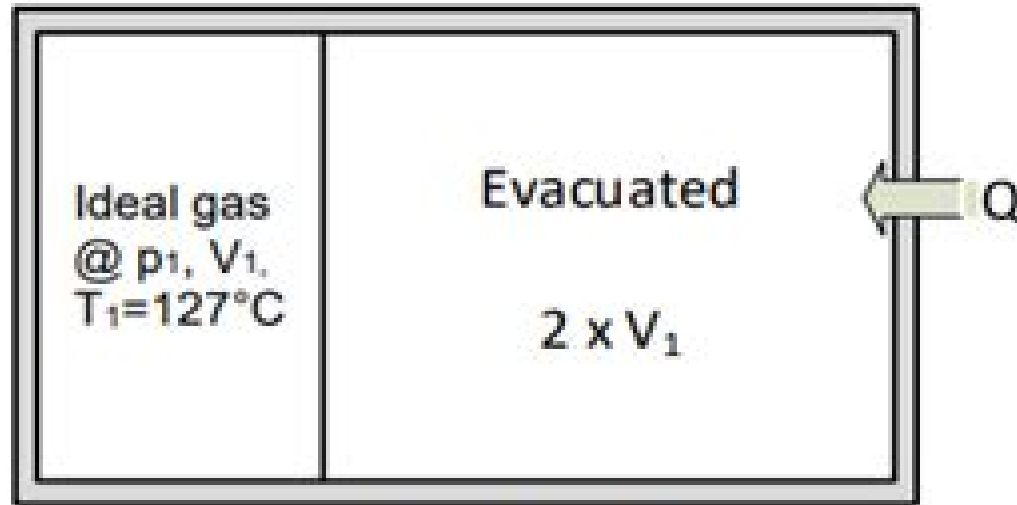


Fig. Q4 - A rigid tank

- b) During the first expansion process, which property/properties remain unchanged and why? [2 marks]
- c) Determine the pressure, volume and temperature at the end of the expansion process, in relation to the initial state p_1, v_1 and T_1 . [3 marks]
- d) Determine the final temperature of the gas at the end of the heat addition process. [3 marks]

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William Kent



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NBS Monograph ,1963 **Development of High-temperature Strain Gages** J. W. Pitts,Dwight Garrison Moore,1961
A summary is presented of a research program aimed at the improvement of high temperature strain gages of the electrical resistance type Potential ceramic and metal components were evaluated and a gage was devised that was based on these evaluations This gage NBS 5B was flexible and easy to install however it lacked resistance stability at higher temperatures In an attempt to minimize this deficiency ceramic cements were developed that showed greater electrical resistivity than had been previously observed in the range 800 to 1800 degrees Fahrenheit also a technique was devised for increasing the resistance to ground by applying a fired on ceramic coating to the grid of a specifically developed unbacked gage A study was made of the cause of the erratic response of cemented gages that had not been preheated prior to use There were strong indications that the erratic response was caused mostly by the rapid decrease in resistance that accompanied structural changes in the cement *NBS Special Publication* ,1968 Precision Measurement and Calibration: Temperature ,1969
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