

# Fundamentals of Corrosion

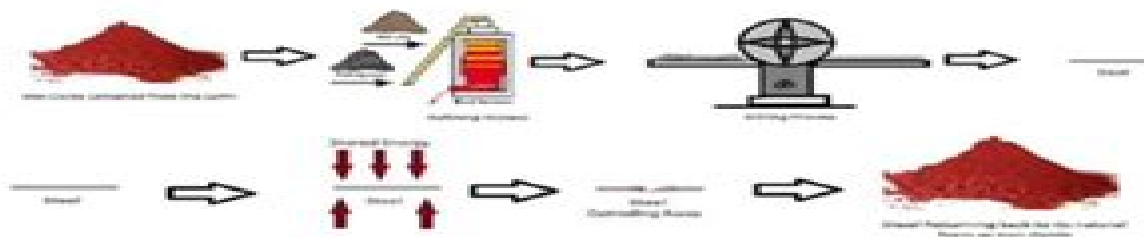
Prof. Dr. Mohsin Ali Raza  
Institute of Metallurgy and Materials Engineering  
University of the Punjab, Lahore, Pakistan

## How Can We Define Corrosion?

- Deterioration of metals due to interaction with environment.
- The tendency of metals to go back to their natural state, Ore.
- Corrosion is an electrochemical deterioration of metals.

### Impact:

5 % of an industrialized nations' income is spent on corrosion prevention and maintenance or replacement of products lost due to corrosion.



### Four essentials for corrosion:

- Anode
- Cathode
- Electrolyte
- Metallic path

Take one away and corrosion is mitigated

## Electrochemical Considerations



Metal undergoing corrosion in an Acid Solution with high concentration of  $H^{+}$  ions



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