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Samuel Johnson

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web solution v1 2 50 l v2 t1 273 30 303 k t2 273 70 343 k we will be applying charles law as pressure is constant here v1 t1 v2 t2 or v2 v1 t1 x t2 2 50 l 303 k x 343 k 2 83 l at 90 c a helium sample has a volume of 500 ml determine the temperature at which the volume of the liquid will become 240 ml