

# **ARNOLD MINDELL SITTING IN THE FIRE**



**Large Group Transformation  
Using Conflict and Diversity**

# Sitting In The Fire Large Group Transformation Using Conflict And Diversity

**Phil Jones**



## **Sitting In The Fire Large Group Transformation Using Conflict And Diversity:**

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web calorimetry worksheet 1 if 0 315 moles of hexane  $\text{C}_6\text{H}_{14}$  is combusted in a bomb calorimeter containing 5 65 liters of water calculate the molar heat of combustion of hexane if the water temperature rises 55 4  $^{\circ}\text{C}$  the specific heat capacity of water is 4 184 J  $\text{g}^{-1} \text{ } ^{\circ}\text{C}^{-1}$   $\Delta H_{\text{comb}} = -4 184 \text{ J } ^{\circ}\text{C}^{-1} \times 55 4 \text{ } ^{\circ}\text{C} \times 5 65 \text{ L} \times 1000 \text{ g } ^{\circ}\text{C}^{-1} = -1310 \text{ kJ}$

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web 6 a 218 g sample of steam at 121 $^{\circ}\text{C}$  is cooled to ice at 14 $^{\circ}\text{C}$  find the change in heat content of the system 7 in going from ice at 34 $^{\circ}\text{C}$  to steam at 138 $^{\circ}\text{C}$  a sample of water absorbs  $1 41 \times 10^5 \text{ J}$  find the mass of the sample answers 1 2 68  $\times 10^4 \text{ J}$  2 6 23  $\times 10^4 \text{ J}$  3 1 71  $\times 10^4 \text{ J}$  4 1 11  $\times 10^5 \text{ J}$  5 2 11  $\times 10^5 \text{ J}$  6

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