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# Solutions Minerals And Equilibria

**Richard A. Sheppard, Arthur James  
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Solutions, Minerals, and Equilibria Robert Minard Garrels, Charles Louis Christ, 1990-01-01      Solutions, Minerals, and Equilibria Robert Minard Garrels, Charles Louis Christ, 1963      Solutions, Minerals, and Equilibria Robert Minard Garrels, 1967      **SOLUTIONS, MINERALS & EQUILIBRIA.** GARRELS., 1954      **Uranium Solution-mineral Equilibria at Low Temperatures with Application to Sedimentary Ore Deposits** Donald Langmuir, 1977      *Thermodynamic Properties of Minerals and Related Substances at 298.15° K (25.0° C) and One Atmosphere (1.013 Bars) Pressure and at Higher Temperatures* Richard A. Robie, David R. Waldbaum, 1968 A summary of the thermodynamic data for minerals at 298 15 K together with calculated values of the functions  $H_0^f$  T  $G_0^f$  T  $S_0^f$  T and  $G_0^f$  T  $H_0^f$  298 5 T at temperatures up to 2 000 K

**PHREEQE** David L. Parkhurst, Donald C. Thorstenson, L. Niel Plummer, 1982      *Thermodynamic Properties of Minerals and Related Substances at 298.150 K (25.00 C) and One Atmosphere (1.013 Bars) Pressure and at Higher Temperatures* Richard A. Robie, David R. Waldbaum, 1968      **Phosphate Minerals** J.O. Nriagu, P.H. Moore, 2012-12-06 The literature on the geology chemistry and biochemistry of phosphorus generally takes its mineralogy for granted The incidental information on phosphate minerals given in these texts is often obsolescent and inaccurate The few mineralogical texts that have dealt comprehensively with the phosphate minerals have now become outdated and typically present the essential information in a manner unsuitable for nongeological readers This volume is intended as a ready reference for workers who require good basic information on phosphate minerals or their synthetic equivalents The topics covered should appeal to geologists and geochemists lithologists environmental scientists and engineers chemists and biochemists who have any interest in the intricate world of phosphorus The hard tissues of many vertebrates and the many pathological calcifications consist mostly of phosphate minerals The precipitation of these compounds also plays a major role in the ecological cycling of phosphorus and occasionally even dominates the behavior of many trace metals in many geochemical and biological systems Indeed many pegmatitic phosphate minerals have acquired some notoriety because of the rarer trace metals which they tend to accumulate With the commercialization of phosphate fertilizers since the early part of the 19th century phosphate minerals have assumed an important role in industrial chemistry and agriculture Clearly the study of phosphate minerals is important from the economic agricultural environmental and human and animal health viewpoint      **Distribution and Genesis of Authigenic Silicate Minerals in Tuffs of Pleistocene Lake Tecopa, Inyo County, California** Richard A. Sheppard, Arthur James Gude, 1968 Zeolites potassium feldspar searlesite and clay minerals formed during diagenesis of rhyolitic vitric tuffs that were deposited in a saline lake      Water-resources Investigations, 1980      *Equilibrium Studies with Certain Acids and Minerals and Their Probable Relation to the Decomposition of Minerals by Bacteria* Douglas Wright, 1922      **Geochemistry and Mineral Formation in the Earth Surface** Rafael Rodríguez-Clemente, Yves Tardy, 1987      Scientific American, 1916      Metamorphic Pressure-temperature-time Paths Frank S. Spear, Simon Muir

Peacock,1989      Soil Solutions, Minerals, and Equilibria H. J. Percival,1985      *Proceedings of the Fourth International Symposium on Electrochemistry in Mineral and Metal Processing* R. Woods,Paul E. Richardson,Fiona M. Doyle,1996      **U.S. Geological Survey Bulletin** Geological Survey (U.S.),1989      *Introduction to Geology and Resources of Gold, and Geochemistry of Gold* Daniel R. Shawe,1988      *U.S. Geological Survey Bulletin* ,1983

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