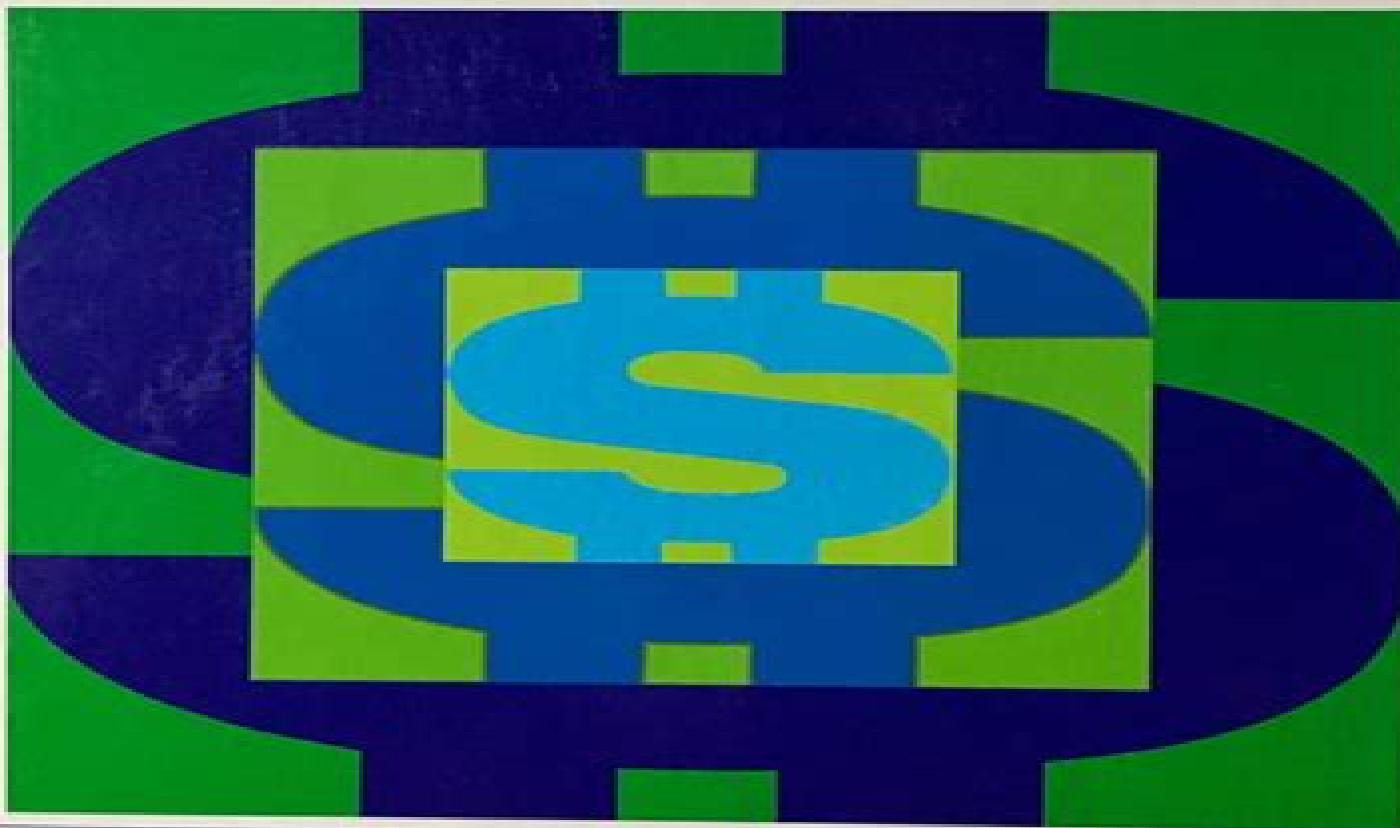


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TWENTY-FOURTH EDITION

Volume 2
General Topics
(Entries 25003-64716)



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Robert R. Redfield



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web ryan is a twelve month old brown haired brown eyed white toddler he is caucasian with almond shaped eyes i met him at a friend s suburban residence on 10 th february 2020 where he and she live i was there between 12 pm and 6 p m at her place i discovered ryan seated on the living room floor

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web may 23 2016 2873 the male child was observed four times over the course of two weeks totaling four hours he was placed in a classroom designated for 2 year olds along with about 12 other children staffed by two full time teachers and a part time teacher the classroom was a very loud and over decorated environment

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web to name just a few examples an educator might use his her observations and analysis to plan an individual intervention with a specific child to make decisions about how best to share observational data with a parent to adjust the furniture in the classroom or to plan the curriculum for the coming week how do early childhood educators observe

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web feb 28 2020 the observation was over and i walked the toddler back to class to play with her friends she ran ahead of me into the room during the observation i always asked if the toddler could do something without me demonstrating with p she could do most of the things with just an audio prop but with l i had to demonstrate every single action i asked

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web apr 1 2015 effective observations will benefit each individual child ensuring that the plans and experiences that follow have been considered based on thorough knowledge of the child and purposeful observations you can find a variety of observation templates and completed samples on our templates page observation templates references

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web sharing observations with families for example you might share how a toddler moved a toy shopping cart back and forth until she figured out how to squeeze it behind a table and you might highlight how the child was learning about space and

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date october 11 child haley mcintire time 9 20 a m 9 30 a m other children and adults present observer child s age mr greene

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web we then changed the volume between 5ml and 20ml and recorded the pressure at each this part of the lab showed that volume and pressure are inversely proportionate as the pressure would decrease if the volume was increased part b of the experiment looked at the relationship between temperature and pressure

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266 section 01 introduction in this lab we observed the relationship between pressure and temperature

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1 10 the capacity of the syringe the 2 0 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law $pV = nRT$ is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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web figure page index 10 since the number of moles in a given volume of gas varies with pressure and temperature changes

chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

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introduction the ideal gas law is a commonly used formula in chemistry it relates pressure temperature and volume this

experiment seeks to prove those relationships by measuring the pressure of gas at various temperatures

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temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of

declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95

82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in

the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

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web figure pageindex 5 the empirically determined relationships among pressure volume temperature and amount of a gas

the thermometer and pressure gauge indicate the temperature and the pressure qualitatively the level in the flask indicates the volume and the number of particles in each flask indicates relative amounts