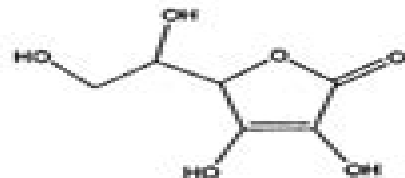
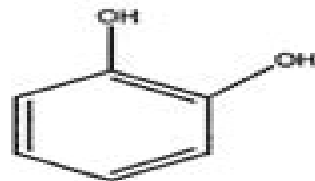


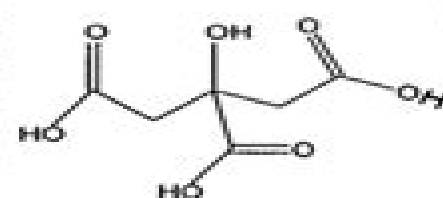
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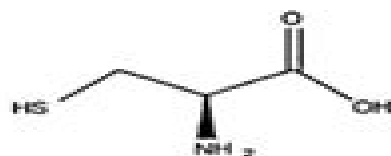
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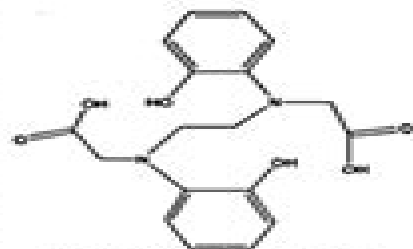
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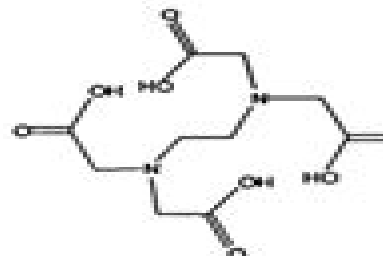
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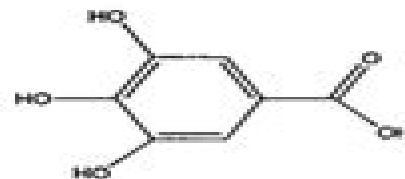
L-cysteine



Ethylenediamine
di(o-hydroxy
phenylacetic)acid
(EDDHA)



Ethylenediamine
tetraacetic acid
(EDTA)



gallic acid



hydroquinone



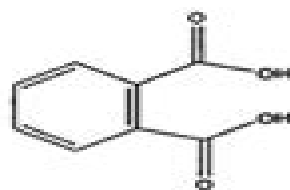
p-hydroxy
benzoic acid



p-nitrophenol



oxalic acid



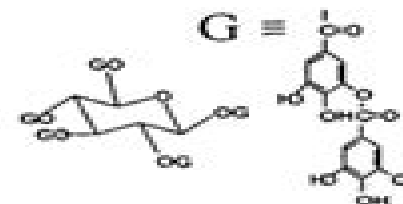
phthalic acid



resorcinol



salicylic acid



tannic acid

Structures Of Organic Molecules

Norman L. Allinger, Janet Allinger



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sulfur and nitrogen derivatives of the above mentioned compounds are also represented with calculated structure dependent energies including alcohols ethers aldehydes and ketones carboxylic acids thiols sulfides amines amides heterocyclic compounds and others Most organic reactions can be interpreted as the disappearance of certain structures and formation of others If the structure dependent energies are known it can be shown how the disappearing and the newly formed structures contribute to the heat of reactions and to the driving forces As experienced by the author who pioneered the concept structure dependent energies can help teachers to make organic chemistry more accessible for their students Accordingly the brief offers a valuable resource for all those who teach organic chemistry at universities and for those who are learning it

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