

What decides the properties of a polymer?

- Stronger attractive forces between chains = stronger, less flexible polymer.
- Chains able to slide past each other = flexible polymer .
- In poly(ethene) attractive forces are weak instantaneous dipole - induced dipole, will it be flexible or not?
- Nylon has strong hydrogen bonds, why does this make it a strong fibre?

Structure And Properties Of Polymers

Rosina Ehmann



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