

Physical Chemistry Formula Revision

Basic Concepts

$$\text{Average atomic mass} = \frac{a_1x_1 + a_2x_2 + a_3x_3 + \dots}{100}$$

 $a_1, a_2, a_3, \dots$ = atomic mass of isotopes. $x_1, x_2, x_3, \dots$ = percentage abundances

How to connect

$$\text{no. of males} = n = \frac{\text{given mass}}{\text{male mass}} = \frac{M}{m}$$

$$n = \frac{\text{no. of particles}}{\text{Avogadro's no.}} = \frac{N}{6.022 \times 10^{23}}$$

$$n = \frac{\text{vol. of gas}}{\text{vol. of gas at STP}} = \frac{V}{22.4 \text{ l}}$$

Concentration Times

$$\text{Percentage, \%} = \frac{W_b \text{ or } V_b}{W \text{ or } V} \times 100$$

$$\text{Molarity, } M = \frac{n_b}{V(l)} = \frac{N_b \times 1000}{M_b \times V(ml)}$$

$$\text{Normality, } N = \frac{n_{\text{eq}}}{V(l)} = \frac{M_b \times 1000}{V_b \times V(\text{ml})}$$
$$\quad \quad \quad \downarrow$$
$$\quad \quad \quad \frac{M_b}{n_f}$$

$$\text{parts per million, ppm} = \frac{W_b}{W} \times 10^6$$

$$\text{Molality, } m = \frac{n_b}{M_b (\text{kg})} = \frac{M_b \times 1000}{M_b \times M_b (\text{g})}$$

$$\text{Mole fraction, } X_b = \frac{n_b}{n_a + n_b + n_c}$$

Stoichiometry



$$\frac{\text{moles of A}}{a} = \frac{\text{moles of B}}{b} = \frac{\text{moles of C}}{c}$$

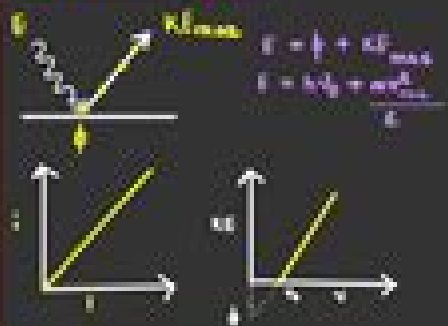
Limiting R_p : $\frac{\text{given amt.}}{\text{stoichiometric coeff.}} \rightarrow \text{least}$

Answer Structure

Plants	Animals	Minerals
--------	---------	----------

$$f = \frac{hc}{\lambda} \quad \text{E (eV)} = \frac{12400}{\lambda (\text{\AA})}$$

Model: Electric Field



The Greenfield's Waypoint Initiative

$$m\lambda = \frac{h_f}{\lambda} \rightarrow \lambda = \frac{h}{mv} \quad \lambda = \frac{h}{\sqrt{2mE}}$$

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$$\text{work} = \frac{m \cdot h}{E \cdot x} \quad \text{with } m = 0.5 \times \frac{m^3}{E} \cdot h^2$$

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$$E_{\text{eff}} = E_0 \left(1 + \frac{E_0}{E_c} \right) \quad \text{with } E_c = \frac{h}{2\pi m^* \lambda_D}$$

$$\frac{1}{\lambda} = \frac{1}{\lambda_0} \left[\frac{1}{\gamma} - \frac{1}{\gamma^3} \right] \quad \gamma = 1.0000000001$$

max. no. of spectral lines when λ^{th} electron from
 n_2 to n_1 : $\frac{(n_2 - n_1)(n_2 - n_1 - 1)}{2}$

Hölder's Inequality Principle: $\|fg\|_1 \leq \|f\|_p \|g\|_q$

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total number = 100

10. **Question:** Product A has a unit volume of 1

$$\text{modified_nodes} = m - 1 - l$$

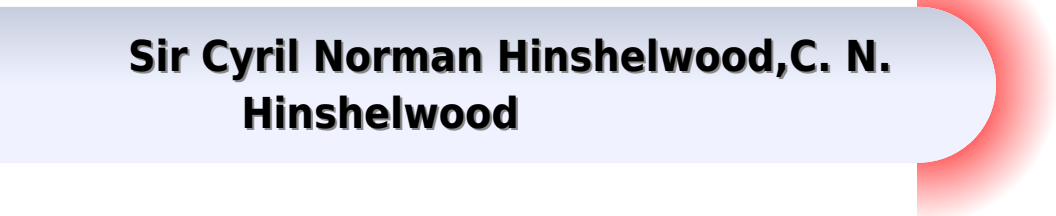
Keywords: *work engagement, organizational commitment, turnover intentions, organizational citizenship behaviors, organizational identification*

Control Group

10. $\frac{1}{2} \times \frac{1}{2} = \frac{1}{4}$

Structure Of Physical Chemistry

**Sir Cyril Norman Hinshelwood, C. N.
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