

Standard Grade Chemistry
Credit Level
Topic 3
Homework 2

1. (2011C)

A student made the following statements about the particles found in an atom.

A	Relative mass = 1
B	Charge = zero
C	Found outside the nucleus
D	Charge = 1+
E	Charge = 1-

Identify the **two** statements which apply to an electron. (1KU)

2. (2010C)

The grid shows information about some particles.

A $^{23}_{11}\text{Na}$	B $^{18}_8\text{O}$	C $^{40}_{19}\text{K}^+$
D $^{24}_{12}\text{Mg}^{2+}$	E $^{35}_{17}\text{Cl}^-$	F $^{16}_8\text{O}$

(a) Identify the **two** particles with the same number of neutrons. (1KU)

(b) Identify the particle which has the same electron arrangement as neon. (1PS)

3. (2009C)

The grid contains information about the particles found in atoms.

A relative mass = 1	B charge = zero	C relative mass almost zero
D charge = 1-	E found outside the nucleus	F charge = 1+

Identify the **two** terms which can be applied to protons. (2KU)

Standard Grade Chemistry With Answers

R.C. SARASWAT



Standard Grade Chemistry With Answers:

Standard Grade Chemistry Norman Conquest, Roddy Renfrew, 2002 The second edition of this title for standard grade chemistry matches the requirements of the standard grade and provides coverage of general and credit level material The authors have revised each chapter taking into account of new developments such as in the field of fossil fuels Other chapters have been updated where relevant for instance to reflect changes to the Relative Atomic Mass data The chemical dictionary chapter has been enlarged to include more key words New exam questions have been included together with homework and whole course questions at the end of the book Answers to Revision Questions for Standard Grade Chemistry John M.

Briggs, 2007 **Standard Grade Chemistry with Answers** Roddy Renfrew, Norman Conquest, 1995 This book has been specifically developed for the new Standard Grade Chemistry syllabus Building on the authors teaching experience the book addresses the real needs of teachers and students and provides a complete coherent course for all students tackling this course and examination

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Christian, Purnendu K. Dasgupta, Kevin A. Schug, 2013-10-07 With the 7th Edition of Analytical Chemistry renowned chemists Purnendu Sandy Dasgupta and Kevin Schug both of the University of Texas Arlington join the author team The new edition focuses on more in depth coverage of the principles and techniques of quantitative analysis and instrumental analysis aka Analytical Chemistry The goal of the text is to provide a foundation of the analytical process tools and computational methods and resources and to illustrate with problems that bring realism to the practice and importance of analytical chemistry It is designed for undergraduate college students majoring in chemistry and in fields related to chemistry **Standard Grade Chemistry** David Hawley, Sandy McLeman, 2012-07-25 1 An easy to use and comprehensive guide to how to achieve the best

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