Activity No.2

SPECIFIC HEAT OF SOLIDS

INTRODUCTION

Calorimetry means "measuring heat." When heat flow occurs between two bodies that are isolated from their surroundings, the amount of heat lost by one body must be equal to the amount of heat gained by the other. Heat is a form of energy in transit, so this principle is just a conservation of energy. We take each quantity added to a body as positive and each quantity leaving a body as negative. When several bodies interact, the algebraic sum of the quantities of heat transferred to all the bodies must be zero.

The mathematical representation of the amount of heat Q needed to achieve a temperature change by a material with mass m and specific heat capacity c is

where T = final temperature, and initial temperature.

In an isolated system, there is no transfer of either mass or energy to its environment. For such case the total heat in the system is conserved, that is,

Specific Heat Capacity for Some Materials/Substance

Material/Substance	Specific Heat c	
	(cal/g C°)	(J/Kg C°)
Water	1.000	4190
Copper	0.093	390
Lead	0.031	130
Steel	0.118	470
Aluminum	0.217	910

OBJECTIVE

To measure the specific heat capacity of two kinds of solid specimens by calorimetry.

APPARATUS

Steam generator with cup, calorimeter, 2 thermometers, 2 kinds of metal shots, platform balance, slotted masses, 600 ml. beaker, electric stove, extension cord, rug

PROCEDURE

https://www.youtube.com/watch?v=d4iCJakz2i8

 Fill the steam generator 1/2 full of water and start heating it over the stove. Set the stove range dial to medium.

Record the two kinds of metal shots in Table 2.1.

3. Weigh about 100 grams of the first metal shots. Pour these in the steam can (inner cup of the steam generator) and place the can in the steam generator. Cover the steam can with a cork with two holes, one for the thermometer and the other for the stirrer. Heat the shots over the steam generator and stir occasionally. Do not use the thermometer to stir the shots but keep the thermometer embedded in the shots. Continue heating and observe the thermometer until all the

Specific Heat Of Solids

Dr. Om Prakash Pahuja

Specific Heat Of Solids:

The book delves into Specific Heat Of Solids. Specific Heat Of Solids is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Specific Heat Of Solids, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Specific Heat Of Solids
 - Chapter 2: Essential Elements of Specific Heat Of Solids
 - Chapter 3: Specific Heat Of Solids in Everyday Life
 - Chapter 4: Specific Heat Of Solids in Specific Contexts
 - ∘ Chapter 5: Conclusion
- 2. In chapter 1, this book will provide an overview of Specific Heat Of Solids. This chapter will explore what Specific Heat Of Solids is, why Specific Heat Of Solids is vital, and how to effectively learn about Specific Heat Of Solids.
- 3. In chapter 2, the author will delve into the foundational concepts of Specific Heat Of Solids. The second chapter will elucidate the essential principles that must be understood to grasp Specific Heat Of Solids in its entirety.
- 4. In chapter 3, the author will examine the practical applications of Specific Heat Of Solids in daily life. This chapter will showcase real-world examples of how Specific Heat Of Solids can be effectively utilized in everyday scenarios.
- 5. In chapter 4, the author will scrutinize the relevance of Specific Heat Of Solids in specific contexts. This chapter will explore how Specific Heat Of Solids is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, the author will draw a conclusion about Specific Heat Of Solids. The final chapter will summarize the key points that have been discussed throughout the book.
 - This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Specific Heat Of Solids.

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