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# Technology Cad Systems Computational Microelectronics

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the oh basic solution one in which the h is less than the oh alkaline solutions the h of a basic solution is less than  $1 \times 10^{-7}$  m

they are the same as basic solutions

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nh<sub>2</sub> and s<sup>2-</sup> are ch<sub>4</sub> nh<sub>3</sub> and hs<sup>-</sup> respectively next we consider that acidity increases with positive charge on the molecule

thus ruling out that s<sup>2-</sup> is the weakest base

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molecule that gains a hydrogen ion becomes this which is positively charged what did lewis propose an acid accepts a pair of

electrons during a reaction while a base donates a pair of electrons

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