

SUPER BABY SONGS™



From the creators of the award-winning, kid-approved BABYSONGS video series, it's the ultimate in toe-tapping tunes for toddlers: SUPER BABY SONGS! With irresistible sing-along songs written by children's favorite HAP PALMER, this joyfully upbeat collection of tunes can be enjoyed on its own, or as the perfect follow-up to the other classic titles in the BABYSONGS series. So slip on your dancin' shoes, strike up the music - and get ready for a singin', laughin', dancin', good time with SUPER BABYSONGS!

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Backwards Land
Chomping Gum
When Things Don't Go
Your Way
Francie Had a Football
When Daddy Was a
Little Boy
If I Had Wings
Hurry Up Blues
Amanda Schlupp
You Can Do It
Teddy Bear Ball

SUPER BABYSONGS
was formerly titled:
TURN ON THE MUSIC

Color/Approx.
Running Time
30 Min.

Duplicated in SP Mode

**VIDEO™
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© 1993 Backyard Enterprises, Inc.
Super BabySongs/Turn On The
Music is produced by Amy Westraub
and Ronald McQueen. Baby Songs is a
trademark of Backyard Productions.
Songs and lyrics written and performed by Hap
Palmer. The songs on this program are from the
"Backwards Land" and "Baby Zoo" videocassettes from Hap-Pal Music, Inc.
Box 325, Trumbull, CA 92788

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Super Baby Songs

D Keegan



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g}) \quad \Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g})$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l}) \quad \Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l}) \quad \Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = 285 \text{ kJ/mole}$

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web thus for reaction 9 2 in which 1 mole of h₃o and 1 mole of oh combine to form 2 moles of h₂o h 55 8 kj as shown note that the enthalpy change h h_{products} h_{reactants} is positive if heat is absorbed that is if h_{products}

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