

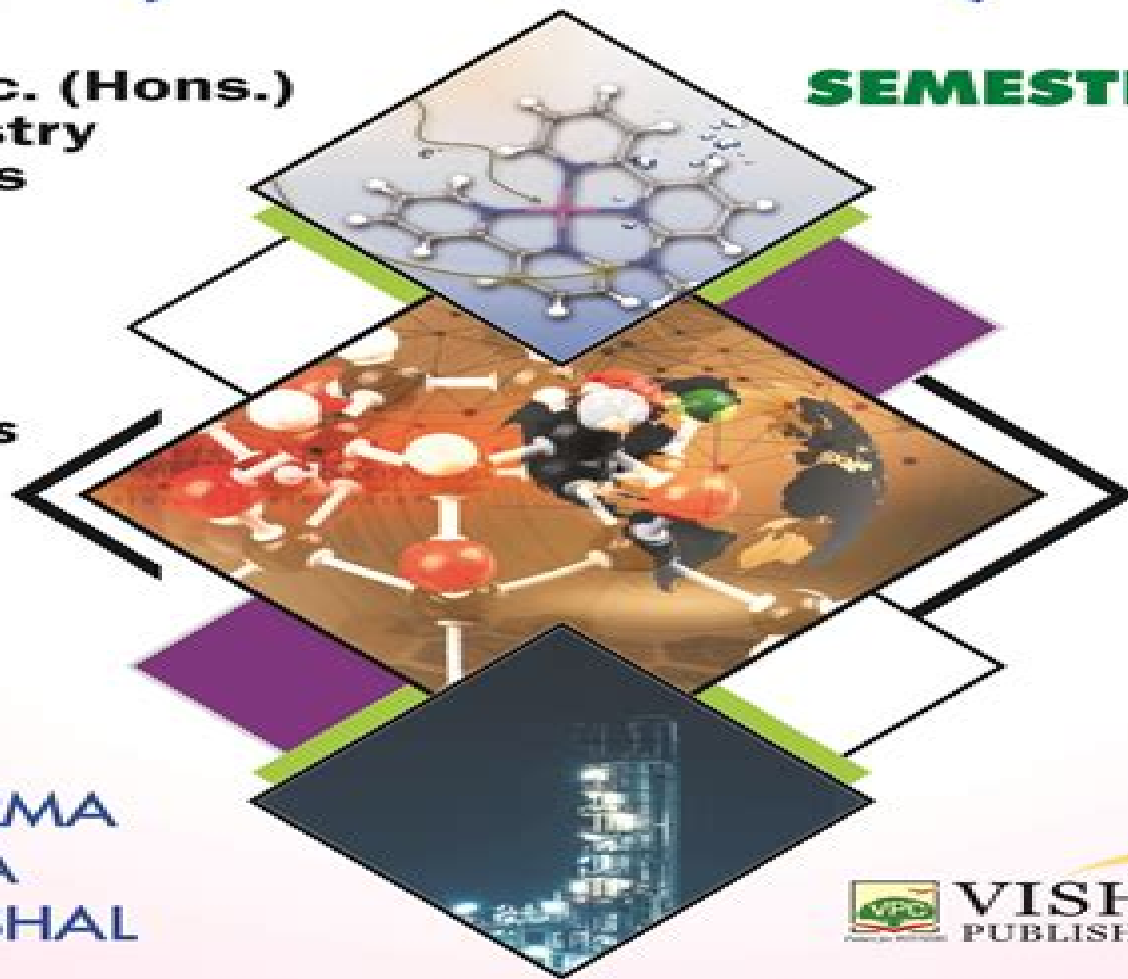
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Textbook Of Inorganic Chemistry

Christian Drosten



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Principles of Inorganic Chemistry Brian W. Pfennig, 2015-03-03 Aimed at senior undergraduates and first year graduate students this book offers a principles based approach to inorganic chemistry that unlike other texts uses chemical applications of group theory and molecular orbital theory throughout as an underlying framework This highly physical approach allows students to derive the greatest benefit of topics such as molecular orbital acid base theory band theory of solids and inorganic photochemistry to name a few Takes a principles based group and molecular orbital theory approach to inorganic chemistry The first inorganic chemistry textbook to provide a thorough treatment of group theory a topic usually relegated to only one or two chapters of texts giving it only a cursory overview Covers atomic and molecular term symbols symmetry coordinates in vibrational spectroscopy using the projection operator method polyatomic MO theory band theory and Tanabe Sugano diagrams Includes a heavy dose of group theory in the primary inorganic textbook most of the pedagogical benefits of integration and reinforcement of this material in the treatment of other topics such as frontier MO acid base theory band theory of solids inorganic photochemistry the Jahn Teller effect and Wade's rules are fully realized Very physical in nature compare to other textbooks in the field taking the time to go through mathematical derivations and to compare and contrast different theories of bonding in order to allow for a more rigorous treatment of their application to molecular structure bonding and spectroscopy Informal and engaging writing style worked examples throughout the text unanswered problems in every chapter contains a generous use of informative colorful illustrations

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A Text-book of Inorganic Chemistry J. R. Partington, 1953 **Inorganic Chemistry**, 2012 **A Textbook of Inorganic Chemistry** A K De, 1969 **Inorganic Chemistry** James E. House, 2012-12-31 Inorganic Chemistry Second Edition provides essential information for students of inorganic chemistry or for chemists pursuing self study The presentation of topics is made with an effort to be clear and concise so that the book is portable and user friendly The text emphasizes fundamental principles including molecular structure acid base chemistry coordination chemistry ligand field theory and solid state chemistry It is organized into five major themes structure condensed phases solution chemistry main group and coordination compounds with several chapters in each There is a logical progression from atomic structure to

molecular structure to properties of substances based on molecular structures to behavior of solids etc The textbook contains a balance of topics in theoretical and descriptive chemistry For example the hard soft interaction principle is used to explain hydrogen bond strengths strengths of acids and bases stability of coordination compounds etc Discussion of elements begins with survey chapters focused on the main groups while later chapters cover the elements in greater detail Each chapter opens with narrative introductions and includes figures tables and end of chapter problem sets This new edition features new and improved illustrations including symmetry and 3D molecular orbital representations expanded coverage of spectroscopy instrumental techniques organometallic and bio inorganic chemistry and more in text worked out examples to encourage active learning and to prepare students for their exams This text is ideal for advanced undergraduate and graduate level students enrolled in the Inorganic Chemistry course This core course serves Chemistry and other science majors The book may also be suitable for biochemistry medicinal chemistry and other professionals who wish to learn more about this subject area Concise coverage maximizes student understanding and minimizes the inclusion of details students are unlikely to use Discussion of elements begins with survey chapters focused on the main groups while later chapters cover the elements in greater detail Each chapter opens with narrative introductions and includes figures tables and end of chapter problem sets

Inorganic Chemistry Gary Wulfsberg, 2000-03-16 Both elementary inorganic reaction chemistry and more advanced inorganic theories are presented in this one textbook while showing the relationships between the two *A Textbook Of Inorganic Chemistry* Arnold Frederick Holleman, 2023-07-18 *A Textbook of Inorganic Chemistry* Richard Ernst Wilhelm Sommer, 2015 *A Text-Book of Inorganic Chemistry* Newth George Samuel, 2013-06 Unlike some other reproductions of classic texts 1 We have not used OCR Optical Character Recognition as this leads to bad quality books with introduced typos 2 In books where there are images such as portraits maps sketches etc We have endeavoured to keep the quality of these images so they represent accurately the original artefact Although occasionally there may be certain imperfections with these old texts we feel they deserve to be made available for future generations to enjoy **TEXTBOOK OF INORGANIC CHEMISTRY (LARGE TEXT CLASSIC REPRINT).** RICHARD ERNST WILHELM. SOMMER, 2018 **A Textbook of Inorganic Chemistry** Richard E. W. Sommer, 1906 *A Textbook of Inorganic Chemistry - Volume 1* Mandeep Dalal, 2017-01-01 An advanced level textbook of inorganic chemistry for the graduate B Sc and postgraduate M Sc students of Indian and foreign universities This book is a part of four volume series entitled *A Textbook of Inorganic Chemistry* Volume I II III IV CONTENTS Chapter 1 Stereochemistry and Bonding in Main Group Compounds VSEPR theory d p bonds Bent rule and energetic of hybridization Chapter 2 Metal Ligand Equilibria in Solution Stepwise and overall formation constants and their interactions Trends in stepwise constants Factors affecting stability of metal complexes with reference to the nature of metal ion and ligand Chelate effect and its thermodynamic origin Determination of binary formation constants by pH metry and spectrophotometry Chapter 3 Reaction Mechanism of Transition Metal Complexes I Inert and labile complexes

Mechanisms for ligand replacement reactions Formation of complexes from aquo ions Ligand displacement reactions in octahedral complexes acid hydrolysis base hydrolysis Racemization of tris chelate complexes Electrophilic attack on ligands
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