



# Slow Hands

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web 1 how does atomic radius change as you move from left to right across a period explain your answer in terms of the number of protons in the nucleus 2 how does atomic radius change as you move from top to bottom in a group explain your answer in terms of the number of electron shells 3 on the representation of the periodic table below

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down a group the outermost electrons are in successively more extensive orbitals as  $n$  increases

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