



## Evidence supporting the proposed mechanism

- The introduction of radical scavengers did not inhibit the reaction, suggesting that the free radicals are not major intermediates.
- Cu(II) complex was characterized by EPR and UV-Vis spectra when *NH*-sulfilimine was subjected to the CuBr solution, supporting that *NH*-sulfilimine could facilitate the oxidation of Cu(I) to Cu(II).
- The  $H_2$  gas was detected by headspace gas chromatography (GC) analysis and the hydrogen atom origin of  $H_2$  was confirmed to be from alcoholic solvent by mass spectroscopy while  $CD_3OD$  was employed as the solvent.
- The copper hydride pathway was plausible based on a density functional theory (DFT) computational study.



# Sulfilimines And Related Derivatives

**Michael Charles Matelich**





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