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web log K_a 10⁻⁴ 0.58 91 x 10⁻⁵ mol dm⁻³ log pK_a pH 3.75 4.05 0.30 10.0 30.0 50 and $HCOOH$ 0.1 so $HCOO^-$ 0.200 mol dm⁻³ amount vol x

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