Dissolving Salt in Water

VALID ARGUMENT FOR BOTH SIDES

Chemical Change

Dissolving salt may be written as a chemical equation. New products are formed, so a chemical reaction occurs.

 $NaCl(s) \rightarrow Na^{+}(aq) + Cl^{-}(aq)$



Physical Change

Salt doesn't react
with the solvent
(water), so no
chemical reaction
occurs. If you remove
the solvent, the
original compound
remains.

Slow Dissolve

Susanne Gannon, Ampersand Pasley, Jayne Osgood



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