

Unit 1**1A Personality****Exercise 1** (page 3)

- 1 shy 2 nasty 3 serious 4 stupid 5 weak 6 polite
7 generous 8 talkative 9 patient 10 honest 11 arrogant

Exercise 2 (page 3)

The opposite of hard-working is lazy.

Exercise 3 (page 3)

- 1 dishonest 2 generous 3 rude 4 talkative 5 clever
6 nasty 7 lazy 8 funny 9 impatient

Exercise 4 (page 3) **LISTENING 2** (page 4)

1. Terry: I love going out with my friends. Terry: He always pays for everybody. He just enjoys spending money! He's so different from his brother. That's my dad of course! The only problem with Harry is that he can be rather impolite. He isn't very sensitive and doesn't wait before he speaks. The other day he really upset my dad when he told him he was mean.
 2. Julie: My mum's amazing, really. She never loses her temper, she just stays really calm. She's got four children. Mine, my brother and my two sisters and that's all who do things but my sister, before we actually do them - but she doesn't get angry. She also works full time at the supermarket and doesn't get home till 10 o'clock. I don't know how she fits it all in. She never stops?
 3. Maria: My sister and I have completely different personalities. I like meeting new people and am happy to chat to people. I don't know - at parties, for example. But Sonia's a bit nervous in situations like that. She waits for people to come to her. But when she gets to know you and starts talking she really makes you laugh.
 4. Teresa: We're twins, but we aren't identical. We don't look alike, and our personalities aren't alike, either. For example, I don't say very much, normally I'm fairly quiet. Louise is the complete opposite. She never stops! But she doesn't talk about herself. Just to me - for example she got fantastic marks in her exams but you never hear her talking about it.
- 1 generous, rude 2 patient, hard-working
3 shy, funny 4 talkative, modest

Challenge! (page 3) **LISTENING 2** (page 4)

1. We can be rather impolite.
2. She never loses her temper, she just stays really calm.
3. Sonia's a bit nervous in situations like that.
4. I'm fairly quiet, Louise is the complete opposite.
- 1 rather impolite
- 2 her temper, really calm
- 3 left, bit nervous
- 4 fairly quiet, complete opposite

Exercise 5 (page 3)

- 2 irresponsible 3 insensitive 4 unfair 5 irrational
6 immature 7 intolerant 8 disloyal

Exercise 6 (page 3)

- 1 sensitive 2 fit 3 immature 4 irrational 5 tolerant
6 disloyal 7 fair 8 irresponsible

1B Present simple and continuous**Exercise 1** (page 3)

- 2 ✓ 3 ✓ 4 I'm wearing trainers today. 5 ✓
6 Look! That man is wearing pink shoes.
7 'What does Sam do?' 'He's a dentist!'

Exercise 2 (page 3)

- 1 Am, having 2 I'm shopping 3 are, looking for 4 never wear
5 like 6 know 7 do, need
8 I'm going

Exercise 3 (page 3)

- 1 a Am, coming b do, come
2 a wash b I'm wearing
3 a Does, see b Are, taking
4 a plays b is playing
5 a hate b I'm enjoying

Exercise 4 (page 3)

- 1 are, doing 2 I'm texting 3 he going 4 are, laughing
5 is going 6 are, doing 7 I'm going 8 isn't going

Exercise 5 (page 3)

- 1 a 2 a 3 b 4 a 5 b 6 b 7 b 8 b

Challenge! (page 3)

Students' own answers

1C Lifestyles**Exercise 1** (page 3)

Students' own answers

Exercise 2 (page 3)

- 1 G 2 F 3 C 4 E 5 D 6 B 7 A

Exercise 3 (page 3)

hospitable: friendly and welcoming
fertile: good soil or land
independent: free from outside control
patriotic: supporting one's country
sensible: based on good judgement
vibrant: full of energy and life
weaving: making a fabric
woodcarving: making an object from wood
ceramics: pots made from clay
city dwellers: people who live in the city
hanging out: spending time relaxing
achievements: things done successfully
prominent: important or famous

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atrium to those in formation we model in catechist formation what we can but in the end it is the catechist who experiences
the atrium with the children who will have to figure out the art of introducing the atrium to the children

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web the catechesis of the good shepherd is an approach to the religious formation of children ages 3 to 12 it originated in
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web the three components of the program the catechist the space and the materials provide a partial answer the catechist
usually there is more than one adult with the children and the children are in multi age groups ages 3 6 6 9 and 9 12 the
session lasts two hours unfolding as follows

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web with a strong acid and strong base titration strong acids and strong bases refer to species that completely dissociate to form ions in solution by contrast weak acids and bases ionize only partially and the ionization reaction is reversible

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web aug 30 2022 acid base titrations are usually used to find the amount of a known acidic or basic substance through acid base reactions the analyte titrand is the solution with an unknown molarity the reagent titrant is the solution with a known molarity that will react with the analyte

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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the titration of hydrochloric acid a strong acid with sodium hydroxide a strong base and acetic acid a

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1:1
 $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
 one mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide NaOH and oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web acid base reactions can be used to determine concentrations of an acid or a base using stoichiometric calculations and titrimetric analysis the volume of a solution required to react

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web Jul 19 2023 introduction titration is an analytical quantitative technique used to determine the concentration of a solute a pH titration is used to determine the concentration of an acid or a base titrations play an important role in determining amount and purity in many manufacturing processes

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being NaOH reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4 + 2\text{NaOH(aq)} \rightarrow \text{Na}_2\text{C}_2\text{O}_4 + 2\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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web conclusion in this lab we used titration to explore the concepts of stoichiometry and equivalence points we found the concentration of an unknown substance by mixing 2m hcl with the naoh of unknown concentration in order to experimentally ascertain the concentration of the naoh

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction $\text{NaOH} + \text{C}_6\text{H}_4\text{COOH} \rightarrow \text{C}_6\text{H}_4\text{COONa} + \text{H}_2\text{O}$ the point of equivalence is the point of titration when the number of moles of NaOH added is equal to the number of moles of KHP in the