



Some people think they need more to be happy. But the truly happy people are those who learn to enjoy life with exactly what they have at each moment.

The Happy Life

Nils Salzgeber



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temperature change plugging in the values given in the problem you get $q_{\text{water}} = 4.18 \text{ J/g} \cdot \text{C} \times 110 \text{ g} \times 26.6 \text{ C} = 12100 \text{ J}$
 $550 \text{ J} = \Delta h \cdot q_{\text{water}} = 550 \text{ J}$

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 for water $4.184 \text{ J/g} \cdot \text{C}$ 1 what is the specific heat of aluminum if the temperature of a 28.4 g sample of aluminum is increased by 8.1 C when 207 J of heat is added 2

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 $14.7 \text{ J} = 3.14 \Delta H_{\text{rxn}} = n \cdot q_{\text{rxn}} = n \cdot q_{\text{calorimeter}} = m \cdot c \cdot \Delta T$

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