



Figure 1. Schematic representation of the diffusion and boundary layers at the electrode-solution interface as applied in the diffusion layer model. Reprinted from [20] by permission of John Wiley & Sons, Inc.

Increased hardness of the deposit, enlarged film thickness, improved deposition rates and efficiencies, greater adhesion of the deposits to electrode, and brightness enhancement are the most often mentioned benefits of electrodeposition of metals assisted with ultrasound [21–23]. The brightness of deposits depends on the current density and it decreases when current density increases above a critical value. The use of ultrasound increases the critical current density and helps to obtain electrodeposits with increased brightness. The ultrasound effects that influence the brightness come from shock waves followed by cavitation erosion. As a result, films built of fine grain size and higher grain-packing density are produced.

The synergism between sound and light in sonophotochemical processes is most often correlated with the enhanced mass transfer induced by ultrasound that can affect the photochemical reaction rates [6]. However, the interaction between sono- and photochemical irradiation strictly depends on the type of ultrasound effect produced in the liquid, which is shown in detail in Figure 2. It can be seen that this enhanced mass transfer can in consequence improve light absorption and modify the reaction pathway. The energy transfer to molecules occurs only when cavitation is produced. The transfer of photoexcited states in a pure photochemical system is not homogeneous (see Figure 3a), because it depends on how light propagates through an absorbing medium and normally its intensity decreases with a distance [24]. It is also known, that the reaction rate in photochemical process depends on the concentration of the excited states. Figure 3b predicts how the distribution of a photoexcited state could change when sonication is added to the system. An intense mixing provoked by cavitation and shock waves should distribute the excited states in the whole reaction volume causing the reaction rates to be more or less the same at each point of the solution. However, not only cavitation is the important factor, but also the relative formation rates, lifetimes, and further decomposition of excited species. Additionally, it was suggested that long-lived excited states could be quenched by cavitation. Another aspect of “sonophoto” synergy can be found in some works where it is said that the combined effects of ultrasound and light could potentially generate a new chemistry of photoexcited solutes under the extreme conditions inside cavitation microreactors [6]. The Colmenares group performed an experiment comparing the deposition of iron on TiO_2 semiconductor, in order to obtain iron oxide, using ultrasound, light, and ultrasound coupled with light (results not published). They compared the morphology and chemical structure of these three samples to find out if a new

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