

Sterile processing practice final exam 2022/2023

Which of the following best describes the purpose of evenly distributing large sets of instruments?

- A. The set is difficult to carry if the weight is unevenly distributed
- B. Extra condensation will make the set more difficult to dry
- C. It is easier for the end user to sort the instruments
- D. The overall weight of the set is affected. - ANSWER-Extra condensation will make the set more difficult to dry.

Which of the following best describes the overall role of the sterile processing area? To provide

- A. Dependable service to enhance patient care
- B. sterilization services on a rigid schedule
- C. Delivery of facility supplies
- D. Support of the operating room technicians - ANSWER-Dependable services to enhance patient care

Which of the following government agencies has the responsibility for approving new environmental disinfectants prior to their being sold to healthcare facilities?

- A. Occupational safety and health administration
- B. Food and drug administration
- C. environmental protection agency
- D. Center for disease control and prevention - ANSWER-Environmental protection agency

The function of the ratchet can be checked by

- A. Closing the ratchet all the way then tapping the handle on the work table
- B. Closing the ratchet on the second click then tapping the handle on the work table
- C. Leaving the ratchet open and moving the jaws of the instrument
- D. Closing the ratchet on the first click then tapping the handle on the work table. - ANSWER-Closing the ratchet on the first click then tapping the handle on the work table

The science that studies the structure of the body is known as

- A. Anatomy
- B. Physiology
- C. Body basics
- D. Microbiology - ANSWER-Anatomy

Which of the following transports molecules within the cell?

- A. Cilia
- B. Flagella
- C. Nucleus

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G Psacharopoulos



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web answers 1 hc 2h 3o 2 l 2o 2 g co 2 g 2h 2o l hc 2h 3o 2 l 4o 2 g 871 7 kj 2co 2 g 2h 2o l 2 the enthalpy of reaction is h
prod h react the reactant is at a higher potential than the product when the reaction occurs heat is released stabilizing the
system and the system moves to a lower energy potential 3

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web thermochemical equations quiz this online quiz is intended to give you extra practice in performing thermochemical
calculations with a variety of reactions including limiting reagents and percent yield options this quiz aligns with the
following ngss standard s hs ps1 4 hs ps3 1

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web in this set of practice questions we will summarize the main concepts of thermochemistry such as the relationship
between internal energy work and heat exothermic and endothermic process heat capacity constant pressure calorimetry
constant volume calorimetry the enthalpy the standard enthalpies of formation and their use in