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Superconducting Properties of BiPbSrCsCuO and BiSrCaCuO Glass-Ceramics

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In this work we study the main differences between the superconducting properties of the $\text{Bi}_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_x$ and $(\text{Bi}_{0.8}\text{Pb}_{0.2})_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_x$ glass-ceramics. The $\text{Bi}_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_x$ and $(\text{Bi}_{0.8}\text{Pb}_{0.2})_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_x$ glass-ceramics prepared in the same conditions contain similar amounts of the superconducting 2212 phase, however their superconducting and normal-state electrical properties differ significantly. The main reason which makes BiSrCaCuO glass-ceramics worse superconductor than BiPbSrCaCuO is its microstructure.

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1. Introduction

Oxide $\text{Bi}_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_x$ and $(\text{Bi}_{0.8}\text{Pb}_{0.2})_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_x$ glasses, annealed above 470°C, crystallise. The crystalline phases forming during annealing are, among others, oxide superconductors: $(\text{Bi,Pb})_2\text{Sr}_2\text{CuO}_x$ (2201 with $T_c = 10$ K), $(\text{Bi,Pb})_2\text{Sr}_2\text{CaCu}_2\text{O}_x$ (2212, $T_c = 85$ K), and $(\text{Bi,Pb})_2\text{Sr}_2\text{Ca}_2\text{Cu}_2\text{O}_x$ (2223, $T_c = 105$ K) [1]. Glass-ceramic are granular materials which usually show a double superconducting transition. At higher temperature isolated grains transit into the superconducting state, while at lower one the grains couple into a long-range ordered state [2]. In this work we study the main differences between the superconducting properties, especially transition temperatures of the $\text{Bi}_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_x$ and $(\text{Bi}_{0.8}\text{Pb}_{0.2})_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_x$ glass-ceramics.

2. Experimental

The $\text{Bi}_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_x$ and $(\text{Bi}_{0.8}\text{Pb}_{0.2})_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_x$ glasses were prepared from nitrates and oxides. The substrates, after calcination at 820°C for 10 h, were

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