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bonds an enolate anion is formed starting aldol condensation the enolate anion comes from an aldehyde or a ketone enolizable aldehydes and ketones undergo a reaction when either an acid or base catalyst is present

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web an aldol reaction was performed by combining acetone and benzaldehyde to make dibenzalacetone the percent yield of dibenzalacetone was found to be 73 which is a good percent yield the melting point of dibenzalacetone was around the range of 104 110 c this helped conclude that dibenzalacetone had a trans trans e e isomer

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